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An investigation of (Na_{0.5}K_{0.5})NbO₃–CaTiO₃ based lead-free ceramics and surface acoustic wave devices

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Abstract

Lead-free $(Na_{0.5}K_{0.5})NbO_3$ ceramics doped with CaTiO₃ (0–3 mol%) have been prepared by the conventional mixed oxide method in this paper. All of the CaTiO₃ doped $(Na_{0.5}K_{0.5})NbO_3$ specimens do not deliquesce as exposed to water for a long time. The samples are characterized by X-ray diffraction analysis, Raman scattering spectra, scanning electron microscopy, and atomic force microscopy. The dielectric, piezoelectric and ferroelectric properties are also investigated. The results show that the addition of CaTiO₃ is very effective in preventing the deliquescence and in improving the electric properties of $(Na_{0.5}K_{0.5})NbO_3$ ceramics. Finally, surface acoustic wave devices based on lead-free ceramics have been successfully fabricated and their characterization is presented.

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1. Introduction

Single crystals and thin films for piezoelectric applications are mostly produced by special techniques, requiring increased process optimization and high cost. Modified ceramics have good potential due to their lower processing cost and ability to facilitate a desirable combination of properties, such as high surface phase velocity, electromechanical coupling coefficient (k^2) and low temperature coefficient of frequency (TCF). Lead oxide (PbO) based ceramics have been widely investigated and used for transducers, piezoelectric actuators, surface acoustic wave (SAW) filters and sensors because of their excellent piezoelectric properties.¹ We have reported many researches about modified lead titanate (PT) and lead zirconate titanate (PZT) ceramics used for SAW applications.^{2,3} However, high volatilization of PbO may drive non-stoichiometry during firing process and its toxicity can contaminate the environment and damage human health. With the raise of environmental consciousness, therefore, it is the present tendency to develop excellent lead-free materials replacing Pb-based piezoelectric ceramics.

Sodium potassium niobate ($(Na_vK_{1-v})NbO_3$, NKN) ceramic is an attractive material and has been thoroughly investigated as a result of its high k^2 and high phase transition temperature ($T_c \sim 420$ °C), especially near the morphotropic phase boundary (MPB).⁴⁻⁶ Nevertheless, dense NKN ceramics are difficultly obtained since their phase stability is limited to 1140 °C.⁷ Moreover, they would deliquesce once exposed to humidity due to the formation of extra phases.^{6,7} The main problem is the volatilization of potassium oxide (K₂O) at 800 °C making the stoichiometry difficult to control.^{7,8} Oxygen deficiency has been another problem in the preparation, which results from high-temperature processing and gives rise to electronic conductivity.⁸ Many researchers used hot pressing or spark plasma sintering (SPS) techniques to yield better quality ceramics.⁹ Recently, an efficient solution to improve foregoing problems is realized by utilizing some additives in NKN ceramics, such as Pb₃O₄, PbTiO₃, ZnO, SrTiO₃ (ST), BaTiO₃, LiNbO₃, LiTaO₃, etc.^{7,10–17} The comparison of properties of

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NKN ceramics based on the previous reports of various groups are demonstrated in Table 1. Furthermore, several groups have studied the effects of 0.5 mol% alkaline-earth (AE) dopant (Mg²⁺, Ca²⁺, Sr²⁺, Ba²⁺) and 0.5 mol% AETiO₃ on the properties of NKN-based ceramics.^{18,19} It has been indicated that CaTiO₃ doping can promote densification of NKN ceramics and improve their properties.¹⁸

At present, the aim of this work is not only to synthesize $(Na_{0.5}K_{0.5})NbO_3$ -CaTiO₃ (NKN-CT) ceramics by the conventional mixed oxide method and investigate their properties, but also to fabricate surface acoustic wave (SAW) devices further.

2. Experimental

The raw materials of $[(Na_{0.5}K_{0.5})_{1-x}Ca_x](Nb_{1-x}Ti_x)O_3$ samples (NKN_{1-x}CT_x), where $x = 0-3 \mod \%$, processed by a conventional mixed oxide method were pure reagent Na₂CO₃, K₂CO₃, Nb₂O₅, CaCO₃ and TiO₂ powders (>99.0% purity). First, 50 g of Na₂CO₃, K₂CO₃ and Nb₂O₅ powders were ballmilled in a 1000 mL polyethylene jar for 12 h with 400 g of 10 mm diameter ZrO₂ balls using 350 mL of ethanol (99.5% purity) as medium. The same procedure was applied to ball mill CaCO₃ and TiO₂ powders. Then, these slurries were separately dried and calcined at 950 °C in air for 10 h, respectively. After pulverization, the two powder batches were weighed according to the stoichiometric formula and ball-milled together. The resulting slurry was then dried, calcined and pulverized sequentially. These powders, milled with 8 wt.% PVA aqueous solution (5%), were uni-axially pressed into a disk of 12 mm diameter, at pressure of 25 kg/cm² and subsequently sintered in air at 1040–1120 °C, depending on the CaTiO₃ contents. A K₂O-rich atmosphere was maintained with NKN-CT powder to minimize the potassium loss during sintering process.

Bulk densities were measured by the Archimedes method using distilled water as medium. The crystallographic study was confirmed by X-ray diffraction (XRD) using Cu Ka $(\lambda = 0.154 \text{ nm})$ radiation with a Seimens D-5000 diffractometer operated at 40 kV and 40 mA. The Raman scattering spectra excited by the 532.25 nm radiation were obtained with a micro-Raman spectrometer (Jobin Yvon, Labram HR) in the $200-1000 \,\mathrm{cm}^{-1}$ range at room temperature. The microstructure was observed by field emission scanning electron microscopy (FESEM) with a Hitachi S-4100 microscope. The dielectric (measured from 60 to 510 °C at 10 kHz) and piezoelectric properties were measured with a HP 4294A precision impedance analyzer. To measure the electrical properties, silver paste was painted on both sides of the samples to form electrodes, and then subsequently fired at 120 °C for 10 min. After that, samples were poled under 40 kV/cm DC field at 150 °C in a silicone oil bath for 30 min. The electromechanical coupling factor in thickness (k_t) and planar (k_p) mode was calculated from the resonance-antiresonance method. Ferroelectric hystersis loops (P-E) were obtained under 50 kV/cm ac field at 60 Hz by a modified Sawyer-Tower circuit.²⁰ The samples were also submerged in 150 °C silicon oil to prevent arcing.

In order to fabricate the SAW devices, the pellets were polished to a mirror finish on one side. Then, the interdigital

Comparison of properti-	es of Na _{0.5} K ₀	1.5 NbO3 cei	ramics t	based on the	previous 1	reports of v	various groups							
Ceramic composition	ρ (g/cm ³)	$\rho_{\rm r}$ (%)	ε,	ε'' (%)	$k_{\rm t}$ (%)	$k_{\rm p}$ (%)	d33 (pC/N)	Nt (kHz-mm)	N _p (kHz-mm)	Q_{m}	TCF _B (ppm/°C)	E _c (kV/cm)	$P_{\rm r}~(\mu{\rm C/cm^2})$	Reference
(hot pressed)	4.46	98.9	I			45	160	I	I	240	I	1	33	6,7
JKN	4.25	94.2	290	I	I	36	80	I	I	130	I	I	I	6,12
JKN	4.3	95.3	472	I	45	39	110	I	I	I	I	20	20	6
VKN-ZnO	~ 4.26	94.5	500	I	I	40	123	I	I	~ 200	I	12	16	12
VKN-SrTiO ₃	4.44	98.4	412	4.0	43.8	32.5	96	I	I	I	I	I	I	7
VKN-BaTiO ₃	4.44	98.4	I	3.9	38	29	104	I	I	I	I	12	7.5	15
VKN-LiNbO3	4.35	96.5	Ι	Ι	48	44	235	I	I	I	I	I	I	16
VKN-LiTaO ₃	I	I	570	I	I	36	200	I	I	I	I	12.5	6	17
VKN-CaTiO ₃	4.40	97.6	553	4.0	42.1	37.9	115	2618	3377	267	-311	12.2	12.4	Our sample

transducer (IDT) electrodes, made of 150 nm aluminum by thermal evaporation method, were patterned onto the polished surface using the lift-off photolithographic process. The surface roughness of polished ceramics was examined by atomic force microscopy (AFM) with a NT-MDT P7LS microscope using the tapping mode. The device characterization was measured with a HP 8714ES network analyzer (Agilent Technologies, Palo Alto, CA).

3. Results and discussion

It is commonly known that adding AE to substitute the perovskite structure by creating A site vacancies in case of NKN leads to an increase in density due to enhancement of sintering mechanism.¹⁸ Adding CT and ST to NKN also can promote densification.¹⁹ Fig. 1 shows the bulk densities of NKN_{1-x}CT_x ceramics and their relative densities. The bulk density at 0.5 mol% CT addition reaches up to 4.40 g/cm³, where the theoretical density is 4.51 g/cm³.⁶ The densities of the rest of the doped ceramics are in the range of 4.33–4.38 g/cm³, equivalent to the relative densities 96–97%. None of the doped samples exhibit deliquescence against water, indicating that the CT addition can reduce the formation of unstable secondary phases enhancing the stability.

The X-ray diffraction patterns of the NKN_{1-x}CT_x ceramics are shown in Fig. 2. The XRD pattern of CaTiO₃ powder is also added for comparison. The NKN–CT ceramics are composed of orthorhombic NKN and cubic CT. The main identified phase matches the orthorhombic NKN with space group *Amm*2 at room temperature, and the results reveal that the orthorhombic perovskite structure is preserved, when x is varied from 0 to 3%.²¹ 2 θ values for the orientations as a function of CT dopant are listed in Table 2. As increasing the CT content, the peaks shift toward a higher angle because the host Na⁺ (ionic radius 1.39 Å) and K⁺ (1.64 Å) are replaced by doped Ca²⁺ (0.99 Å) and Nb⁵⁺ (0.64 Å) is replaced by Ti⁴⁺ (0.61 Å), respectively.^{13,22}

XRD confirms the macroscopic symmetry in long-range ordering and Raman spectroscopy is sensitive to non-uniform distortions of the crystal lattice in short-range ordering.²¹ According to the literature, the vibrations of NbO₆ octa-



Fig. 1. The bulk densities of $NKN_{1-x}CT_x$ ceramics and their relative densities.



Fig. 2. X-ray diffraction patterns of the NKN $_{1-x}$ CT $_x$ ceramics.

Table 2 2θ values for the orientations as a function of CT dopant

CT dopant (mol%)	2θ value for the orientation (°)			
	(100)	(110)	(200)	
0	22.250	31.885	45.362	
0.25	22.242	31.873	45.345	
0.5	22.268	31.908	45.395	
1	22.275	31.891	45.406	
2	22.279	31.894	45.438	
3	22.277	31.896	45.450	

hedra are made up of $1A_{1g}(v_1) + 1E_g(v_2) + 2F_{1u}(v_3,v_4) + F_{2g}(v_5) + F_{2u}(v_6)$ in this case, where v means the frequency band and A_{1g} , E_g , F_{1u} , F_{1u} , F_{2g} , and F_{2u} indicate six species of vibration modes).^{21,23} Of these vibrations, $1A_{1g}(v_1) + 1E_g(v_2) + 1F_{1u}(v_3)$ are stretching and the rest are bending modes.²¹ Fig. 3 shows the Raman spectra of NKN_{1-x}CT_x ceramics measured at room temperature. v_5 (262 cm⁻¹) and v_1 (616 cm⁻¹) are detected as relatively strong scattering. As the CT content increasing, the peaks for v_1 shift to a higher frequency and the full-width at half-maximum



Fig. 3. Raman spectra of NKN_{1-x}CT_x ceramics measured at room temperature.

(FWHM) also becomes broader. It is because of the incorporation of $CaTiO_3$ into the perovskite structure, which increases the binding strength of O–Nb–O.

Fig. 4 shows the SEM images and the grain size distributions of $NKN_{1-x}CT_x$ ceramics. It is distinctly observed that the grain size becomes smaller gradually with the increase of the CT content and thus, the doped ceramics are denser than the pure NKN. From the corresponding distributions of the grain size, it is very

clear to note the increase in the number of small grains by adding CT.

Antiferroelectric NaNbO₃ possesses many phase transitions in wide temperature range (-100-643 °C). However, ferroelectric KNbO₃ combined with NaNbO₃ produces a new ferroelectric phase and displays three phase transitions around -120, 200 and 420 °C in the whole temperature range, corresponding to the transition temperatures of rhombohedral to



Fig. 4. The SEM images and the grain size distributions of $NKN_{1-x}CT_x$ ceramics.

orthorhombic, orthorhombic to tetragonal and tetragonal (ferroelectric) to cubic (paraelectric), respectively.^{4,12,24} Fig. 5 shows the dielectric constant as a function of the temperature, and the dielectric constant and loss tangent versus CT content at room temperature. In Fig. 5(a), two-phase transitions are observed obviously above the room temperature and the peaks become broader as the increase of CT content. The two transition temperatures and the maximum value of dielectric constant at these temperatures also decrease with increasing CT additives. Previous investigations reported that the phenomenon has been found in many compounds,^{7,13} such as (Pb,La)(Zr,Ti)O₃ and doped BaTiO₃. It is suggested that a transition from a normal ferroelectric to 'relaxor-like' ferroelectric due to the cation disorder in perovskite unit cell and the formation of microdomain since non-ferroelectric materials are added. In the experiment, the value of the Curie temperature for CaTiO₃ additive is found to be negative,²² which exhibits quantum paraelectric phase at room temperature or higher. The relaxor behavior can be produced by many reasons.^{7,13,14} In Fig. 5(b), the dielectric constant measured at room temperature and 10 kHz increases with increasing CT content, and the entire loss tangent of CT doped NKN is around 4%, which is less than pure NKN. Furthermore, according to the experimental results, the dielectric constant and loss tangent decrease with increasing the measuring frequency. Although their frequency dispersion both exist clearly between 1 and 10 kHz, the apparent higher



Fig. 5. (a) The dielectric constant as a function of the temperature for $NKN_{1-x}CT_x$ ceramics at 10 kHz and (b) the dielectric constant and loss tangent vs. CT content at room temperature.



Fig. 6. Dependence of k_t and k_p of the NKN ceramics on the CT content.

frequency dispersion (10 kHz, 100 kHz, and 1 MHz) are not oberved. 19

Dependence of k_t and k_p of the NKN ceramics on the CT content is shown in Fig. 6. CT doped ceramics (0.5 mol%) exhibit the maximum value of k_t and k_p , reaching 42.1 and 37.9%, respectively. Addition of small amounts of CT yields to larger electromechanical coupling factors than those of pure NKN samples ($k_t = 39.8\%$ and $k_p = 37.6\%$). The promotion may be attributed to the increased density, lowering the leakage current and enhancing the poling process.¹⁵ However, when the CT content exceeds 0.5 mol%, the values of k_t and k_p decrease gradually since the CT with a symmetry structure has no piezoelectricity. Other variations of piezoelectric properties are shown in Fig. 7, including the piezoelectric constant (d_{33}) , the thickness frequency constant (N_t) , planar frequency constant (N_p) , mechanical quality factor (Q_m) and TCF_B (the subscript "B" means bulk) values. The experimental results show that the NKN ceramics with 0.5 mol% CT provide the maximum values of $d_{33} = 115$, $N_t = 2618$ kHz-mm, $N_p = 3377$ kHz-mm and $Q_{\rm m}$ = 267. The ceramics with higher frequency constant can be used in higher frequency applications at identical bulk volume, and those with higher $Q_{\rm m}$ provide reduced the mechanical loss. In addition, TCF_B of all NKN_{1-x}CT_x ceramics are negative and their absolute values increase correspondingly with increasing CT additive, which were determined from the shift of center frequency measurements at temperature from 25 to 80 °C, using the following equation:

$$\Gamma CF_{\rm B} = \frac{f(80\,^{\circ}{\rm C}) - f(25\,^{\circ}{\rm C})}{(80 - 25)f(25\,^{\circ}{\rm C})}$$

Fig. 8 shows the ferroelectric properties of the NKN_{1-x}CT_x ceramics measured at 150 °C and 60 Hz. The remanent polarization (P_r) is 15.7 μ C/cm² and the coercive electric field (E_c) is 15.9 kV/cm for pure NKN ceramics. The P_r and E_c of 0.5 mol% CT content are 12.4 μ C/cm² and 12.2 kV/cm, and those for 2 mol% CT are 8.4 μ C/cm² and 10.1 kV/cm, respectively. With further increasing the doping content of CT, the values of P_r and E_c decrease. The results are shown in Fig. 8(b). The decrease in P_r might be ascribed to the non-ferroelectric CT at room temperature, whereas, the decrease in E_c is due to the increase of the bulk density that



Fig. 7. Variations in d_{33} , N_t , N_p , Q_m and TCF_B values of the NKN_{1-x}CT_x ceramics.

diminishes the leakage current enhancing the polarization in process.^{12,15}

Surface acoustic wave can be generated at the free surface of an elastic solid. In the SAW devices, the generation of the waves is achieved by application of a voltage to a metal-film IDT deposited on the surface of a piezoelectric solid. Two IDTs are required in the basic SAW device. One of these acts as the device input and converts signal voltage variations into mechanical acoustic wave. The other IDT is employed as an output receiver to convert mechanical SAW vibrations back into output voltages. Such energy conversions require the IDTs to be used in conjunction with elastic surfaces that are also piezoelectric ones.²⁵ A SAW device configuration with IDTs/NKN-CT ceramic structure is shown in Fig. 9. IDT patterns were fabricated on the polished ceramic surface. The input and output IDTs consisted of 15.5 finger pairs with 20 µm width of electrode and 20 µm separation, i.e., a periodicity of 80 µm. The IDT center-to-center separation was 2.84 mm with a 4 mm aperture. Fig. 10 shows the frequency response of the SAW device for NKN_{1-x}CT_x ceramics with x = 0.5%. The center frequency of 40.475 MHz leads to a phase velocity of 3238 m/s, which is higher than many commercialized polycrystal-ceramics, for example, Pz27 (2016 m/s) and Pz34 (2510 m/s).²⁶ The insertion loss is quite large around 36.9 dB. One reason is owing to



Fig. 8. (a) The *P*–*E* hysteresis loops and (b) P_r and E_c variations of the NKN_{1-x}CT_x ceramics measured at 150 °C and 60 Hz.

the rough surface probably. Fig. 11 shows the AFM images of $NKN_{0.995}CT_{0.005}$ ceramics with a polished surface. There are many cavities or voids on the surface. The root-mean-square roughness (Rq) without and with a polished surface are, respectively, 165.65 and 8.0 nm. The other reason may be due to itself



Fig. 9. A SAW device configuration with IDTs/NKN-CT ceramic structure.



Fig. 10. The frequency response of the SAW device for NKN_{1-x}CT_x ceramics with x = 0.5%.



Fig. 11. The AFM images of $NKN_{0.995}CT_{0.005}$ ceramics with a polished surface.

characteristics. Moreover, it has high $k^2 \sim 5.37\%$ obtained from the equation²⁶:

$$k^2 = \left(\frac{\pi}{4N}\frac{G_{\rm a}}{B}\right)_{f=f_{\rm c}}$$

where *N* is the number of IDT fingers, and G_a and *B* are radiation resistance and susceptance, respectively. The k^2 is higher than most of PT-based ceramics (~1–4%).²⁶ Its TCF_S (the subscript "S" means SAW) was calculated by substitution of center frequency for 0, 20, and 40 °C into the following equation:

$$TCF_{S} = \frac{f(40 \,^{\circ}C) - f(0 \,^{\circ}C)}{(40 - 0)f(40 \,^{\circ}C)}$$

Table 3 The parameters of NKN–CT ceramics and SAW device properties

NKN–CT ceramic		
CT (mol%)	0.5	
Sintering temperature (°C)	1100	
Sintering time (h)	3	
Poling filed (kV/cm)	40	
Poling time (min)	30	
SAW device		
f(MHz)	40.48	
v (m/s)	3238	
Insertion loss (dB)	-36.92	
k^2 (%)	5.37	
TCF _S (ppm/°C)	-292	

The value arrives about $-292 \text{ ppm/}^{\circ}\text{C}$, which is higher than that of the SAW device based on single-piezocrystal substrate, such as LN128 (+75 ppm/ $^{\circ}$ C), LN64 (-81 ppm/ $^{\circ}$ C) and LT36 (-32 ppm/ $^{\circ}$ C).²⁵ It is also much higher than Pb-based ceramic substrate (\sim -120–60 ppm/ $^{\circ}$ C).^{2,3,25,26} The parameters of NKN–CT ceramics and SAW device properties are listed in Table 3.

4. Conclusion

The properties of NKN_{1-x}CT_x ceramics with x = 0-3 mol%by the conventional mixed oxide method have been investigated. None of the CT doped NKN specimens deliquesce as exposed to water for a long time. The dielectric constant at room temperature and the number of small grain sizes increases, and the transition temperatures decrease with increasing of CT content. CT doped NKN ceramics (0.5 mol%) show excellent piezoelectric properties of $k_t = 42.1\%$, $k_p = 37.9\%$, $d_{33} = 115$, $N_{\rm t} = 2618 \,\text{kHz-mm}, N_{\rm p} = 3377 \,\text{kHz-mm}$ and $Q_{\rm m} = 267 \,\text{which are}$ better than pure NKN. However, the absolute values of the bulk TCF increase correspondingly with increasing CT additive. SAW devices based on NKN_{1-x}CT_x (x = 0.5 mol%) ceramics have been successfully fabricated. The devices perform a phase velocity of 3238 m/s, high k^2 of 5.37%, and TCF about $-292 \text{ ppm/}^{\circ}\text{C}$. High k^2 and velocity make it possible to fabricate electromechanical transducers, however, it is unsuitable for applications on SAW filters ascribe to too large negative TCF. This unfavorable condition may have born fruitful results by doping or coating a thin film on it. As such, it might be promising for sensor applications because the high value of k^2 and velocity can enhance the sensitivity. Also the slight variation of temperature would be easily observed with high TCF value.

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